

Mid-Infrared Emission Characteristic and Energy Transfer of Ho³⁺-Doped Tellurite Glass Sensitized by Tm³⁺

G. X. Chen · Q. Y. Zhang · G. F. Yang · Z. H. Jiang

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Abstract We report on 2.0- μm emission characteristic and energy transfer of Ho³⁺-doped tellurite glass sensitized by Tm³⁺ upon excitation of a conventional 808 nm laser diode. The Judd-Ofelt strength parameters, spontaneous radiative transition probabilities and radiative lifetime of Ho³⁺ have been calculated from the absorption spectra by using the Judd-Ofelt theory. Significant enhancement of 2.0- μm emission of Ho³⁺ has been observed with increasing Tm³⁺ doping up to 0.7 mol%. The energy transfer coefficient of the forward Tm³⁺ \rightarrow Ho³⁺ is approximately 17 times larger than that of the backward Tm³⁺ \leftarrow Ho³⁺ energy transfer. Our result indicates that the maximum gain of 2.0- μm emission, assigned to the transition of ⁵I₇ \rightarrow ⁵I₈ of Ho³⁺, might be achieved from the tellurite glass at the concentration of 0.5 mol% of Tm₂O₃ and 0.15 mol% of Ho₂O₃. The high gain coefficient and quantum efficiency (1.16) along with the large value of the product of the stimulated emission cross-section and the measured radiative lifetime ($4.12 \times 10^{-27} \text{ m}^2\text{s}$) of the Ho³⁺/Tm³⁺-codoped tellurite glasses might find potential applications in efficient 2.0- μm laser.

Keywords Mid-infrared laser · Fluorescence · Energy transfer · Gain · Tellurite glass

Introduction

The mid-infrared (mid-IR) lasers operating at the range of 2.0–3.0 μm and beyond provide potential for application

in laser medicine surgery, remote chemical sensing, eye-safe laser radar, the monitoring of atmospheric pollutions, high resolution spectroscopy of low pressure gasses [1–6]. Ever since Johnson [7] has reported the first laser action at 2.0 μm from Ho³⁺ doped CaWO₄ crystal in 1962, an intense level of research activity has been focusing on measuring the spectroscopic and lasing properties of Ho³⁺ doped various crystals and glasses [8–10].

Tellurite glasses exhibit high linear and nonlinear refractive indices, good resistance to corrosion, high mechanical stability and low cut-off phonon energy of around 750 cm^{-1} . Moreover, the transmission range of tellurite glasses extends into the mid-IR up to about 5 μm and it has good capacity to accept lanthanide dopants. The low phonon energy and wide transmission range of tellurite glasses allow the observation of rare-earth ions doped lasers emissions in a large optical range. These features, coupled with good resistance to corrosion and high mechanical stability, make tellurite glasses become promising host for mid-IR lasers at 2.0- μm wavelength.

In this article, we report on 2.0- μm emission characteristic and energy transfer of Ho³⁺-doped TeO₂-WO₃-ZnO tellurite glass sensitized by Tm³⁺ upon excitation of a conventional 808 nm laser diode (LD). Effect of Tm³⁺ dopant on the fluorescence properties of 2.0- μm wavelength from Ho³⁺ has been investigated.

Experimental procedures

TeO₂-based glasses with the molar compositions of 70TeO₂-20WO₃-10ZnO codoped with 0.3 mol% Ho₂O₃ and x mol% Tm₂O₃ ($x = 0, 0.1, 0.3, 0.5$ and 0.7), and 70TeO₂-20WO₃-10ZnO codoped with 0.5 mol% Tm₂O₃ and y mol% Ho₂O₃ ($y = 0, 0.01, 0.05, 0.10$ and 0.15) were prepared by

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using reagent-grade TeO_2 (99.99%), WO_3 (99.99%), ZnO (99.99%), Ho_2O_3 (99.99%) and Tm_2O_3 (99.99%). Batch materials of 15 g were mixed well and melted at about 850°C for 20 minute with a platinum crucible. Melts were thermally quenched by casting the melt into a preheated stainless steel mold, and then annealed at a temperature close to the vitreous transition temperature for two hours before ramping down to room temperature. All the obtained samples were cut into specimens of $10\text{ mm} \times 10\text{ mm} \times 2\text{ mm}$ and optically polished for the measurement of the absorption and emission spectra.

Absorption spectra of rare-earth doped samples were determined by a Perkin-Elmer Lambda 900/UV/VIS/NIR spectrophotometer in the spectral range of 400–2300 nm at room temperature. The fluorescence spectra in the range of 1600–2200 nm were obtained by using a computer-controlled TRIAX 320 fluorescence spectrometer (Jobin-Yvnon Corp.) with a 808 nm LD as pump source.

Results and discussion

Spectroscopy characteristics and Judd-Ofelt analysis

Figure 1 shows the room temperature absorption spectra in the UV-visible and IR regions of Ho^{3+} -doped tellurite glass sensitized by Tm^{3+} . The absorption bands represent the transitions from the ground state to the excited states of 4f configuration of Tm^{3+} and Ho^{3+} , respectively. Owing to the strong intrinsic bandgap absorption located at 420 nm in the host glasses, absorption from Ho^{3+} cannot be clearly identified below 420 nm. It is worthwhile to mention that the amplitude of the band near 794 nm is no less than those of the other three observed bands of Tm^{3+} , which indicates that Tm^{3+} ion in the glass can be excited efficiently by 808 nm LD. Figure 2 shows the schematic energy levels of Tm^{3+} and

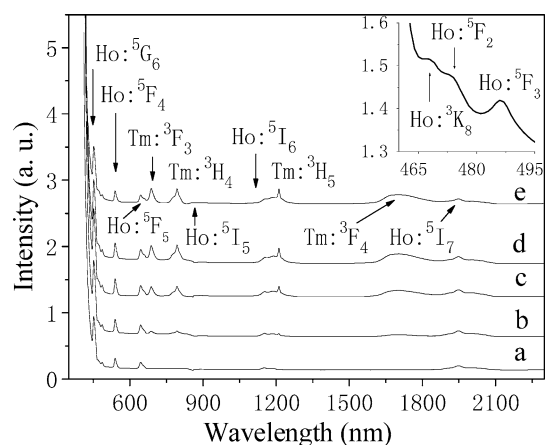


Fig. 1 Absorption spectra of the TeO_2 - WO_3 - ZnO glasses containing (a) 0, (b) 0.1, (c) 0.3, (d) 0.5 and (e) 0.7 mol% Tm_2O_3 with a fixed Ho_2O_3 concentration of 0.3 mol%. Inset shows the amplification of the absorption spectra curves in the range 460–500 nm

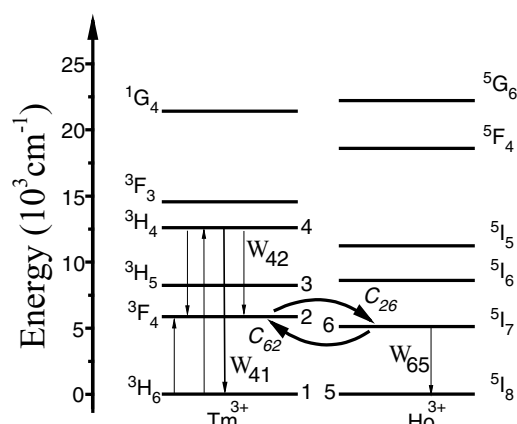


Fig. 2 Schematic energy levels diagram of Tm^{3+} and Ho^{3+} in TeO_2 - WO_3 - ZnO glass under excitation of 808 nm LD

Ho^{3+} identified from the absorption spectra of TeO_2 -based glasses. The locations of the energy levels were very similar to those reported previously for chalcogenide [6], tellurite [11] and fluoride [2] glass hosts.

The Judd-Ofelt analyses [12, 13] were applied using the experimental oscillator strengths of the absorption bands obtained from absorption spectrum. The Judd-Ofelt intensity parameters, Ω_t ($t = 2, 4, 6$), can be figured out utilizing the reduced matrix elements of the unit tensor operators provided by Rukmini et al. [14]. The comparisons of Ω_t in various Ho^{3+} -doped glass hosts are shown in Table 1. Generally, Ω_2 is closely related to the ligand symmetry of host material and the orderliness of structure. It is hypersensitive to the change of compositions in materials. The Ω_t parameters follow the trend $\Omega_2 > \Omega_4 > \Omega_6$ in tellurite glass. Ω_4 , and Ω_6 are related to the covalence of the lattice site of rare earth ions, which can be adjusted by the composition or structure of the glass hosts. The larger the value of Ω_6 , the weaker will be the covalence between rare earth ions and anions. It should be mentioned that the Ω_6 value in tellurite glass (this work) is only less than those of fluorophosphates and ZnBS glass, but

Table 1 The Judd-Ofelt intensity parameters of Ho^{3+} doped various glasses

Glass	Ω_2	Ω_4	Ω_6	Ω_4/Ω_6
ZnBS [14]	22.42	5.19	12.33	0.429
Fluoride [14]	2.43	1.67	1.84	0.908
$\text{PbO-Bi}_2\text{O}_3\text{-Ga}_2\text{O}_3$ [15]	4.77	2.18	1.22	1.7869
Fluorophosphate [2]	2.10	3.5	2.5	1.4
Germinate [2]	3.30	1.14	0.17	0.0671
PbO-SiO_2 [14]	5.20	1.80	1.20	1.5
Phosphate [14]	5.60	2.72	1.87	1.4545
YAlO_3 [16]	1.82	2.38	1.53	1.556
$\text{Ge}_{30}\text{As}_{10}\text{S}_{60}$ [15]	6.98	2.53	0.78	3.2436
NaTeO_2 [16]	6.92	2.81	1.42	1.979
Tellurite ^{This Work}	5.26	2.28	2.18	1.0459

Table 2 Electric dipole linestrengths (S_{ed}), radiative transition probabilities ($A_{ed} + A_{md}$), total radiative transition probabilities (A_r), branching ratios (β) and radiative lifetime (τ_{rad}) for fluorescence levels of Ho^{3+} ions in tellurite glass

SLJ	S'L/J	Energy (cm ⁻¹)	S_{ed} (10 ⁻²⁰ cm ²)	$A_{ed} + A_{md}$ (s ⁻¹)	β
$^5I_7 \rightarrow$	5I_8	5128	3.309064	201 + 56.5	1.00
		$A_r(^5I_7) = 257.5 \text{ s}^{-1} \tau_{rad}(^5I_7) = 3891 \text{ us}$			
$^5I_6 \rightarrow$	5I_8	8620	1.422492	472	0.85
	5I_7	3492	2.253802	50 + 30.43	0.15
		$A_r(^5I_6) = 552.43 \text{ s}^{-1} \tau_{rad}(^5I_6) = 1809 \text{ us}$			
$^5I_5 \rightarrow$	5I_8	11236	0.199865	174	0.39
	5I_7	6108	1.70382	238	0.53
	5I_6	2616	1.773691	19 + 15.034	0.08
		$A_r(^5I_5) = 446.03 \text{ s}^{-1} \tau_{rad}(^5I_5) = 2241 \text{ us}$			
$^5F_5 \rightarrow$	5I_8	15480	2.25566	5128	0.77
	5I_7	10352	1.846296	1255	0.19
	5I_6	6860	1.323036	262	0.04
	5I_5	4244	0.415881	19	0.00
		$A_r(^5F_5) = 6664 \text{ s}^{-1} \tau_{rad}(^5F_5) = 150 \text{ us}$			
$^5F_4 \rightarrow$	5I_8	18553	1.989327	9516	0.81
	5I_7	13425	0.619486	1123	0.09
	5I_6	9933	1.057586	776	0.07
	5I_5	7317	1.239247	364	0.03
	5F_5	3073	1.351096	29 + 10.485	0.00
		$A_r(^5F_4) = 11818.49 \text{ s}^{-1} \tau_{rad}(^5F_4) = 87 \text{ us}$			
$^5F_3 \rightarrow$	5I_8	20576	0.634095	5320	0.48
	5I_7	15448	1.123317	3988	0.36
	5I_6	11956	0.651673	1073	0.10
	5I_5	9340	0.657667	516	0.05
	5F_5	5096	0.598809	76	0.01
	5F_4	2023	0.785002	6 + 5.463	0.00
		$A_r(^5F_3) = 10984.46 \text{ s}^{-1} \tau_{rad}(^5F_3) = 91 \text{ us}$			
$^5G_5 \rightarrow$	5I_8	24200	1.503502	13059	0.40
	5I_7	19042	3.44781	14589	0.45
	5I_6	15510	1.364074	3119	0.68
	5I_5	12932	0.312998	415	0.09
	5F_5	8566	2.202877	849	0.18
	5F_4	5383	1.868582	179	0.04
	5F_3	3431	1.358555	34	0.01
	5F_2	2840	0.448878	6	0.00
	5K_8	2687	0.024813	0	0.00
	5G_6	1721	1.505544	5 + 0.98932	0.00
		$A_r(^5G_5) = 32250.99 \text{ s}^{-1} \tau_{rad}(^5G_5) = 31 \text{ us}$			

larger than those of the others. Since the $^5I_7 \rightarrow ^5I_8$ transition of Ho^{3+} is mainly affected by Ω_6 and A_r (spontaneous emission probabilities) is proportional to Ω_6 [2], the tellurite glass should have high A_r value for the $^5I_7 \rightarrow ^5I_8$ transition of Ho^{3+} . It is interesting to note that the Ω_4/Ω_6 ratio is found to be about one in the glass and fluoride glass. Table 2 represents the (A_r), fluorescence branching ratio (β) and radiative lifetime (τ_r) of different levels of Ho^{3+} ions which were calculated using the calculated Judd-Ofelt intensity parameters. The results show that the values of τ_r and A_r of the 5I_7 level

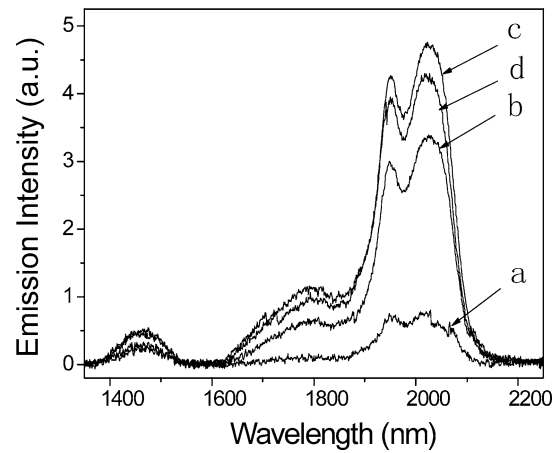


Fig. 3 Fluorescence spectra of the TeO_2 - WO_3 - ZnO glasses containing (a) 0.1, (b) 0.3, (c) 0.5 and (d) 0.7 mol% Tm_2O_3 with a fixed Ho_2O_3 concentration of 0.3 mol% upon excitation at 808 nm LD

of Ho^{3+} are about 3.9 ms and 257.5 s^{-1} , respectively. These values are comparable to that found by Singh [17].

Figure 3 shows the fluorescence spectra of Ho^{3+} -doped tellurite glass sensitized by Tm^{3+} in the range of 1400–2200 nm under excitation of 808 nm LD at the room temperature. The emission peaks at 1.47-, 1.8- and 2.0- μm are assigned to the transitions of Tm^{3+} : $^3H_4 \rightarrow ^3H_6$ and Ho^{3+} : $^5I_7 \rightarrow ^5I_8$, respectively. It can be clearly seen that the emission intensity of the peak at 1.47- μm changes slightly with Tm_2O_3 concentration. The emission intensity of the peak at 1.8- μm increases with increasing Tm_2O_3 concentration while that of the peak at 2.0- μm increases significantly. However, there is no emission band at around 2.0- μm in the Tm_2O_3 singly doped glass. These indicate that the 2.0- μm emission of Ho^{3+} can only originate from energy transfer (ET) of Tm^{3+} to Ho^{3+} ion because Ho^{3+} ion has no absorption band at around 808 nm. Therefore, it is expected that an efficient ET process from Tm^{3+} to Ho^{3+} will play an important role in Tm^{3+}/Ho^{3+} -codoped glasses pumped by 808 nm LD. In addition, a maximum of the emission intensity of 2.0- μm emission is observed for a Tm_2O_3 content of about 0.5 mol%.

Gain spectra characteristics

Figure 4 shows the absorption cross-section and stimulated emission cross-section of Ho^{3+} for the 2.0- μm transition in the TeO_2 - WO_3 - ZnO glasses. The absorption cross-section, $\sigma_a(\lambda)$, was determined from the absorption spectra by using Beer-Lambert equation [3]

$$\sigma_a(\lambda) = \frac{2.303 \log(I_0/I)}{Nl} \quad (1)$$

where I_0 and I are the intensities of incident and transmitted light, respectively, N is the concentration of Ho^{3+} ,

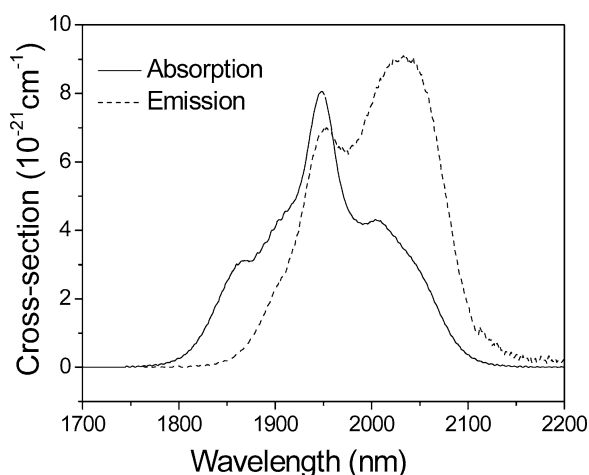


Fig. 4 Absorption cross-sections and the derived emission cross-sections of $\text{TeO}_2\text{-WO}_3\text{-ZnO}$ glasses doped with 0.5 mol% Tm_2O_3 and 0.3 mol% Ho_2O_3

and l is the thickness of the sample. The absorption cross-section of Ho^{3+} : $^5\text{I}_8 \rightarrow ^5\text{I}_7$ transition had a maximum of $8.06 \times 10^{-21} \text{ cm}^2$ at 1948 nm. The stimulated emission cross-section, $\sigma_e(\lambda)$, was calculated from the fluorescence spectra utilizing Fuchtbauer-Ladenburg equation [18]

$$\sigma_e(\lambda) = \frac{\lambda^4}{8\pi c n^2} A_{rg}(\lambda) \quad (2)$$

where $g(\lambda)$ is the normalized line-shaped function obtained from the measured fluorescence spectra, c is the speed of light, n is the refractive index. The emission cross-section of Ho^{3+} : $^5\text{I}_7 \rightarrow ^5\text{I}_8$ transition had a maximum of $9.15 \times 10^{-21} \text{ cm}^2$ at 2027 nm. This value is almost one time larger than that of gallate-bismuth-lead glass ($5.44 \times 10^{-21} \text{ cm}^2$) [3] or fluorite glass ($5.3 \times 10^{-21} \text{ cm}^2$) [2], but less than that of chalcogenide glass ($15.4 \times 10^{-21} \text{ cm}^2$) [6]. The larger absorption and emission cross-section of Ho^{3+} is mainly due to the high refractive index of glass matrix [4] and high spontaneous transition probability. This provides advantages for efficient laser output.

Figure 5 presents the gain spectra of Ho^{3+} near the 2.0- μm wavelength region in $\text{Tm}^{3+}/\text{Ho}^{3+}$ -codoped samples. The calculated gain coefficients could be determined by using: $G(\lambda) = n(^5\text{I}_7)\sigma_e(\lambda) - n(^5\text{I}_8)\sigma_a(\lambda)$ [3], where $n(^5\text{I}_7)$ and $n(^5\text{I}_8)$ are the electron population densities of the $^5\text{I}_7$ and $^5\text{I}_8$ levels in Ho^{3+} , respectively. If electrons in Ho^{3+} are only in either the $^5\text{I}_7$ or $^5\text{I}_8$ state, we can simplify the equation to $G(\lambda) = N[p\sigma_e(\lambda) - (1-p)\sigma_a(\lambda)]$, where N is the concentration of Ho^{3+} , and p is the population inversion rate. The values of $G(\lambda)/N$ could be calculated by assuming different p values. The gain spectra characteristics (Fig. 5) are similar to those of fluorescence spectra in the Ho^{3+} -doped $\text{TeO}_2\text{-WO}_3\text{-ZnO}$ glasses sensitized by Tm^{3+} . The peak wavelength

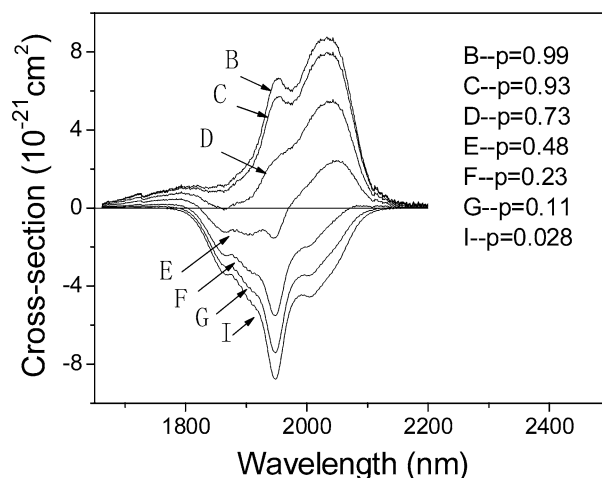


Fig. 5 Gain spectra of Ho^{3+} near the 2.0 μm wavelength region in $\text{Tm}^{3+}/\text{Ho}^{3+}$ -codoped $\text{TeO}_2\text{-WO}_3\text{-ZnO}$ glass

at which the highest gain of 2.0- μm emission occurs is centered at 2027 nm along with a shoulder centered at 1948 nm. When $p > 0.93$, the shape of gain spectrum resembles that of fluorescence spectrum of Ho^{3+} near the 2.0- μm wavelength region. When $0.23 < p < 0.93$, the value of gain near the 1948 nm wavelength region decreases rapidly while that near the 2.0- μm wavelength region is relatively reduced slowly. With the reduction of the value of p (namely, with decreasing with population inversion for $\text{Tm}^{3+}/\text{Ho}^{3+}$ co-doped system), the peak wavelength at which the highest gain of 2.0- μm emission occurs shifts to longer wavelengths. This phenomenon is a typical characteristic of the quasi-three-level laser. Since the gain coefficient is directly proportional to the quantum efficiency of the specific transition (Refs. [3, 19]), indicating the glass studied should have a high quantum efficiency.

Table 3 represents the results of calculated quantum efficiencies (η), peak emission cross-section (σ_e), measured peak wavelength (λ_p) of the emission spectrum and the product ($\sigma_e \times \tau_m$) of the peak emission cross-section and the upper state measured lifetime (τ_m) in various Ho^{3+} -doped glasses. High quantum efficiency is one major parameter of evaluating laser glasses. Quantum efficiency can be expressed as [1, 2] $\eta = \tau_m/\tau_r$. Quantum efficiency of the emitting level is strongly affected by the phonon energy of glass hosts [1, 2]. Generally, the lower the phonon energy of glass hosts, the higher the quantum efficiency of the emitting level. Nevertheless, the fluoride and heavy metal oxides such as gallate-bismuth-lead glass [20] with lower phonon energy have high quantum efficiency. $\sigma_e \times \tau_m$ is another major parameter of evaluating laser glass. Especially, it plays the key role for cw operation [1]. It is interesting to note from Table 3 that the values of the quantum efficiencies in fluoride and tellurite glasses are much higher than those in the other oxide glasses and the value of $\sigma_e \times \tau_m$ in fluoride

Table 3 Spectroscopic properties of various Ho³⁺ doped glasses

	λ_p/nm	$N/\text{wt}\%$	A_r/s^{-1}	τ_m/ms	$\sigma_e/10^{-20}$ cm^{-2}	η	$\sigma_e \times \tau_m/$ $10^{-27} \text{ m}^2\text{s}$
Fluoride [2]	2035	2	58.07	26.7	0.53	1.55	14.15
Gallate [2]	2055	2	69.53	8.2	0.38	0.57	3.12
Fluorophosphate [2]	2035	2	90.42	5.6	0.79	0.651	4.42
Silicate [2]	2040	2	61.65	0.32	0.70	0.02	0.22
Tellurite	2027	2	257.5	4.5	0.915	1.16	4.12

Note. λ_p is measured peak wavelength, N is the dopant concentration, σ_e is the peak emission cross-section.

glass is the largest while that in silicate with high phonon energy is the lowest. The values of $\sigma_e \times \tau_m$ in tellurite, gallate and fluorophosphate glasses are almost equal. Although the measured lifetime (4.5 ms) [21] of the level ⁵I₇ of Ho³⁺ in tellurite glass is less than that in fluoride glass, the values of A_r and σ_e of the level ⁵I₇ → ⁵I₈ transition of Ho³⁺ in tellurite glass are larger than those in fluoride glass. In a word, excluding fluoride glass, tellurite glass is superior to the other oxide glasses according to the values of η and $\sigma_e \times \tau_m$ (see Table 3). Therefore, we expect that tellurite glass will become an important host material of Ho³⁺ doped laser.

Cross-relaxation rate of Tm³⁺ ions and energy transfer analysis

There are at least two energy transfer processes in the Ho³⁺-doped tellurite glass sensitized by Tm³⁺ upon excitation of 808 nm (LD). One is spatial energy transfer among Tm³⁺ ions followed by the cross-relaxation of ³H₆(Tm³⁺) + ³H₄(Tm³⁺) → ³F₄(Tm³⁺) + ³F₄(Tm³⁺) that can improve pump quantum efficiency. Another process is resonant energy transfer from the ³H₄(Tm³⁺) level to the ⁵I₇(Ho³⁺) level that enhances 2.0- μm emission. The cross-relaxation rate (W_{CR}) can be calculated from the measured and calculated fluorescence lifetime of the ³H₄(Tm³⁺) level by using Eq. (3). Because the phonon energy of tellurite glass (~750 cm⁻¹) is small compared to the energy separation of 4360 cm⁻¹ between the level ³H₄ and ³H₅ of Tm³⁺ ions, the multiphonon relaxation rate (W_{MP}) is negligibly small. Then, the cross-relaxation rate can be simplified as Eq. (4).

$$\frac{1}{\tau_m} = \frac{1}{\tau_r} + W_{MP} + W_{CR} \tag{3}$$

$$W_{CR} \cong \frac{1}{\tau_m} - \frac{1}{\tau_r} \tag{4}$$

The result of calculating cross-relaxation rate among Tm³⁺ ions is shown in Table 4.

It is necessary to apply two assumptions to analyze the process of the energy transfer from Tm³⁺ to Ho³⁺ before utilizing a simple model in Fig. 2. First, only the lowest three energy levels of Tm³⁺ beside the ³H₅ level are considered due to the fast multiphonon decay from ³H₅ to ³F₄ level.

Second, because of the low excitation rate from the ⁵I₇ to ⁵F₄, ⁵S₂ level, the upconversion related with the ⁵I₇(Ho³⁺) level is so weak that only the lowest two levels of Ho³⁺ are considered. The rate equations are derived from Shin et al. [3].

$$\begin{aligned} \frac{dn_1}{dt} = & -\phi_{808}\sigma_{Tm}n_1 - Rn_4n_1 + W_{41}n_4 + C_{26}n_2n_5 \\ & - C_{62}n_1n_6 + \frac{n_2}{\tau_2} \end{aligned} \tag{5}$$

$$\frac{dn_2}{dt} = 2Rn_4n_1 + W_{42}n_4 - C_{26}n_2n_5 + C_{62}n_1n_6 - \frac{n_2}{\tau_2} \tag{6}$$

$$\frac{dn_4}{dt} = \phi_{808}\sigma_{Tm}n_1 - Rn_4n_1 - \frac{n_4}{\tau_4} \tag{7}$$

$$n_1 + n_2 + n_4 = n_{Tm} \tag{8}$$

$$\frac{dn_5}{dt} = -C_{26}n_2n_5 + C_{62}n_1n_6 + \frac{n_6}{\tau_6} \tag{9}$$

$$\frac{dn_6}{dt} = C_{26}n_2n_5 - C_{62}n_1n_6 - \frac{n_6}{\tau_6} \tag{10}$$

$$n_5 + n_6 = n_{Ho} \tag{11}$$

where ϕ_{808} and σ_{TM} are the flux of the incident pump photons and the absorption cross-section of Tm³⁺, respectively. C_{ij} and W_{ij} are the energy transfer coefficient and rate from i to j energy level transition, respectively. n_i and τ_i are the electron population and lifetime of the i level, respectively. n_{Tm} and n_{Ho} are the total concentration of Tm³⁺ and Ho³⁺, respectively. $R(W_{CR}/n_{Tm})$ is the cross-relaxation coefficient of

Table 4 Initial conditions used for the calculation of the rate equations (5) to (11)

Parameter	Value	Parameter	Value
ϕ_{808}	1.706×10^{21}	$\sigma_{Tm}(\text{cm}^2)$	6.83×10^{-21}
W_{CR}	4367	$\tau_4(\text{us})$ [22]	294.12
$W_{41}(\text{s}^{-1})$ [22]	3100	$W_{42}(\text{s}^{-1})$ [22]	300

Table 5 The calculated energy transfer coefficients C_{26} and C_{62} in Ho^{3+} -doped tellurite glass sensitized by Tm^{3+} upon excitation of 808 nm laser diode

Tm_2O_3 (mol%)	Ho_2O_3 (mol%)	C_{26} ($\times 10^{-17}$ cm^3/s)	C_{62} ($\times 10^{-17}$ cm^3/s)
0.5	0.15	32.6	1.58
0.5	0.1	31.8	1.63
0.5	0.05	30.8	1.61
0.5	0.01	29.3	1.55

Tm^{3+} . The initial conditions were $n_1(0) = n_{\text{Tm}}$, $n_2(0) = 0$, $n_4(0) = 0$, $n_5(0) = n_{\text{Ho}}$, $n_6(0) = 0$, $\tau_2 = 0.754$ ms [23], $\tau_6 = 3.9$ ms. The other parameters used in the rate equations are also listed in Table 4.

Table 5 shows the forward $\text{Tm}^{3+} \rightarrow \text{Ho}^{3+}$ energy transfer coefficient C_{26} , and vice versa C_{62} in Ho^{3+} -doped samples sensitized by Tm^{3+} upon excitation of 808 nm LD. Energy transfer coefficient C_{ij} can not be directly solved because it is multiplied by the same terms in the simultaneous equations. Another set of equations is needed. It can be obtained by exciting Ho^{3+} from ground state $^5\text{I}_8$ to upper level $^5\text{I}_7$ using a 900 nm excitation source. However, there is a strong energy transfer from Ho^{3+} to Tm^{3+} which is too complicated to analyze. Therefore, energy transfer coefficient C_{ij} will be indirectly solved by utilizing iterative method [3]. From Table 5, we can observe that the forward energy coefficient C_{26} ($\text{Tm}^{3+}: ^3\text{F}_4 \rightarrow \text{Ho}^{3+}: ^5\text{I}_7$) is about 17 times larger than the backward energy coefficient C_{62} ($\text{Tm}^{3+}: ^3\text{F}_4 \leftarrow \text{Ho}^{3+}: ^5\text{I}_7$).

Figure 6 shows the dependence of electron population densities of different levels of Tm^{3+} and Ho^{3+} ions on time in the $\text{TeO}_2\text{-WO}_3\text{-ZnO}$ glasses. Electron population densities of different levels were obtained by numerically solving the rate equations from Table 4, Table 5 and the initial values of different levels described above using the Runge-Kutta method [24, 25]. After exponentially increasing or decreasing for a certain time, the electron population densities of different levels trend to a steady state.

Considering the unchange of the electron population densities of different levels when all energy levels are in steady state, the electron population densities of $^5\text{I}_7$ and $^5\text{I}_8$ levels of Ho^{3+} can be directly read out. The gain coefficients at 2027 nm for tellurite glasses doped with 0.5 mol% Tm_2O_3 and various concentration of Ho_2O_3 were calculated using the gain coefficients expression mentioned above and the results were shown in Fig. 7. Gain coefficient increases with increasing pump power intensity. Although gain coefficient can be obtained with small pump power intensity when the concentration of Ho_2O_3 is low, it increases very slowly with rapidly increasing pump power intensity. On the other hand, gain coefficient can be achieved only at relatively larger pump power intensity when the concentration of Ho_2O_3 is high, but it increases abruptly with increasing pump power

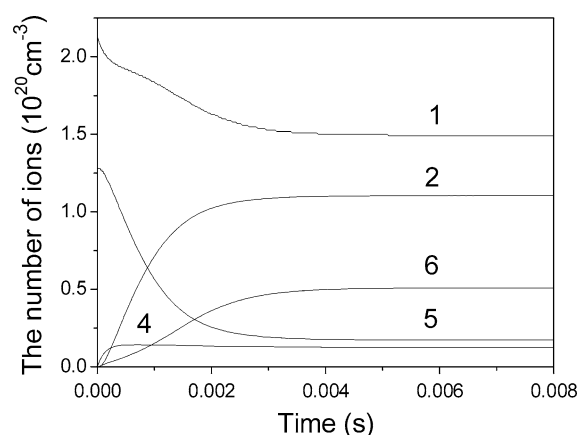


Fig. 6 The dependence of the electron population densities of energy levels of Tm^{3+} and Ho^{3+} on time in $\text{TeO}_2\text{-WO}_3\text{-ZnO}$ glass. The relevant energy levels of Tm^{3+} are (1) $^3\text{H}_6$, (2) $^3\text{F}_4$ and (4) $^3\text{H}_4$ and those of Ho^{3+} are (5) $^5\text{I}_8$ and (6) $^5\text{I}_7$

intensity. The highest gain can be expected from the tellurite glass doped with 0.5 mol% Tm_2O_3 and 0.15 mol% Ho_2O_3 . Comparing gallate glass with the same concentration of Ho_2O_3 , tellurite glass has lower pump power threshold and higher gain coefficient.

Conclusion

In summary, we conclude that 2.0- μm emission characteristic and energy transfer of Ho^{3+} -doped tellurite glass sensitized by Tm^{3+} upon excitation of 808 nm laser diode were investigated. Tm^{3+} ions were codoped to populate the $^5\text{I}_7$ level of Ho^{3+} and resulted in the intense 2.0- μm emission from tellurite glass. The calculated fluorescence lifetime of the $\text{Ho}^{3+}: ^5\text{I}_7$ level was 3.9 ms, and the emission cross section of the $\text{Ho}^{3+}: ^5\text{I}_7 \rightarrow ^5\text{I}_8$ transition exhibited a maximum

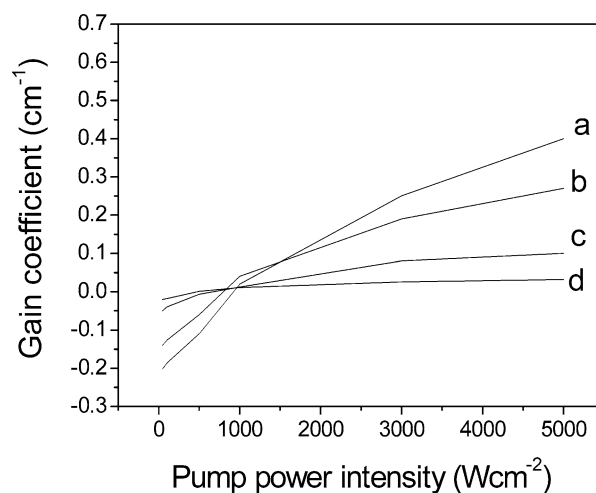


Fig. 7 The dependence of calculated gain coefficients at 2027 nm upon pump power intensity in $\text{TeO}_2\text{-WO}_3\text{-ZnO}$ glasses doped with the Ho_2O_3 concentration of (a) 0.15, (b) 0.1, (c) 0.05 and (d) 0.01 mol% at a fixed Tm_2O_3 concentration of 0.5 mol%

of $9.15 \times 10^{-21} \text{ cm}^2$ at 2027 nm. The coefficients of the forward $\text{Tm}^{3+} \rightarrow \text{Ho}^{3+}$ energy transfer were approximately 17 times larger than those of the backward $\text{Tm}^{3+} \leftarrow \text{Ho}^{3+}$ energy transfer. Our result indicated that the highest gain of 2.0- μm emission, due to the transition of Ho^{3+} : $^5\text{I}_7 \rightarrow ^5\text{I}_8$, might be achieved from the glass at the rare-earth ion concentration of 0.5 mol% of Tm_2O_3 and 0.15 mol% of Ho_2O_3 .

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